

Table 3. Bond lengths and inter-bond angles

C(1)–C(2)	1.389 Å	C(1)–C(2)–C(3)	120.4°
C(2)–C(3)	1.381	C(2)–C(3)–C(4)	119.5
C(3)–C(4)	1.391	C(3)–C(4)–C(5)	119.9
C(4)–C(5)	1.378	C(4)–C(5)–C(6)	120.6
C(5)–C(6)	1.399	C(5)–C(6)–C(1)	119.2
C(6)–C(1)	1.372	C(6)–C(1)–C(2)	120.4
C(1)–O(1)	1.380	C(6)–C(1)–O(1)	122.5
C(2)–O(2)	1.364	C(2)–C(1)–O(1)	117.1
O(1)–H–O(2')	2.795	C(1)–C(2)–O(2)	121.0
O(2)–H–O(1')	2.806	C(3)–C(2)–O(2)	118.6

+0.007, C(4) –0.002, C(5) –0.003, C(6) +0.006 Å. The two oxygen atoms are –0.014 and –0.056 Å respectively out of this plane, this slight distortion from rigorous overall planarity probably being enforced by packing considerations.

The mean C–C bond length is 1.385 Å, and the mean C–O bond length 1.372 Å, which compare with the values of 1.39 and 1.36 Å respectively found in resor-

cinol (Robertson, 1936). The mean hydrogen bond length (2.801 Å) is however rather longer than the mean (2.70 Å) in the resorcinol structure.

References

- BROWN, C. J. (1960). *Acta Cryst.* **13**, 1049.
 CASPARI, W. A. (1926). *J. Chem. Soc.* p. 573.
 CRUICKSHANK, D. W. J. & PILLING, D. E. (1961). In *Computing Methods and the Phase Problem in X-Ray Crystal Analysis*, p. 32. Oxford: Pergamon Press.
 KITAIGORODSKII, A. I. (1947). *Structure Reports*, **11**, 663.
 KITAIGORODSKII, A. I. (1949). *Structure Reports*, **12**, 408.
 LEE, J. D. & WALLWORK, S. C. (1959). *Acta Cryst.* **12**, 210.
 PALIN, D. E. & POWELL, H. M. (1947). *J. Chem. Soc.* p. 208.
 PALIN, D. E. & POWELL, H. M. (1948). *J. Chem. Soc.* pp. 61, 571, 815.
 POWELL, H. M. (1950). *J. Chem. Soc.* pp. 298, 300, 468.
 POWELL, H. M. & RIESZ, P. (1948). *Nature, Lond.* **161**, 52.
 ROBERTSON, J. M. (1936). *Proc. Roy. Soc. A*, **157**, 79.
 ROBERTSON, J. M. & UBBELOHDE, A. R. (1938). *Proc. Roy. Soc. A*, **167**, 122.
 WALLWORK, S. C. (1956). *J. Chem. Soc.* p. 4855.

Short Communications

Contributions intended for publication under this heading should be expressly so marked; they should not exceed about 1000 words; they should be forwarded in the usual way to the appropriate Co-editor; they will be published as speedily as possible. Publication will be quicker if the contributions are without illustrations.

Acta Cryst. (1966). **21**, 174

The space group and unit cell of cyclohex-2-ene-1, 4-dione. By VIRGINIA RYCHNOVSKY, *Department of Chemistry, University of Minnesota, Minneapolis, Minnesota 55455, U.S.A.*

(Received 16 February 1966)

With an interest in the conformation of cyclohex-2-ene-1,4-dione, C₆H₆O₂, prepared by Garbisch (1965), we started the crystal structure determination of this compound. Short needle-shaped crystals were enclosed in capillaries to prevent sublimation. From precession photographs (Mo K α , $\lambda=0.7107$ Å), the monoclinic cell dimensions were found to be

$$\begin{aligned} a &= 6.92 \pm 0.01 \text{ \AA} \\ b &= 6.82 \pm 0.01 \\ c &= 6.07 \pm 0.01 \\ \beta &= 97^\circ 20' \pm 10' \end{aligned}$$

where a is the needle axis. From systematic extinctions ($0k0$, $k=2n+1$; $h0l$, $h+l=2n+1$) the space group was determined to be $P2_1/n$. By comparison with the molecular volumes of quinol (132 Å³/molecule) and quinhydrone (2×126 Å³/molecule), it was concluded that there are two mol-

ecules per unit cell, since this would give a molecular volume for cyclohex-2-ene-1,4-dione of 142.1 Å³. Also, crystals sink very slowly in glycerin, which suggests that the density is slightly greater than 1.28 g.cm⁻³, the value for glycerin. For $Z=2$, the calculated density is 1.285 g.cm⁻³.

With two molecules per unit cell, for space group $P2_1/n$, the molecule must lie on a center of symmetry. Since no reasonable structure for cyclohex-2-ene-1,4-dione contains a center of symmetry, the structure must be disordered. Therefore, we plan no further work on this compound.

The author is grateful to the National Aeronautics and Space Administration for fellowship support.

References

- GARBISCH, E. (1965). *J. Amer. Chem. Soc.* **87**, 4971.